

GENERAL SYNTHESIS OF 2,5-DIHYDROTHIOPHENES (3-THIOLENES) FROM DIKETO SULFIDES

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Abstract: The intramolecular reductive coupling reaction of easily accessible diketo sulfides by a low-valent titanium reagent [prepared from titanium(IV) chloride and zinc powder] provides an efficient general synthesis of 2,5-dihydrothiophenes.

2,5-Dihydrothiophenes are synthetically important compounds and many methods have been devised to prepare this heterocycle.¹ Diketo sulfides are easily accessible compounds in which three functional groups [carbonyl, active methylene (methyne), and sulfide moieties] are suitably arranged for various intra- and intermolecular chemical transformations. However, syntheses with these compounds have not been fully explored. As a part of our synthetic study with diketo sulfides, herein we report a general synthesis of 2,5-dihydrothiophenes from these compounds.

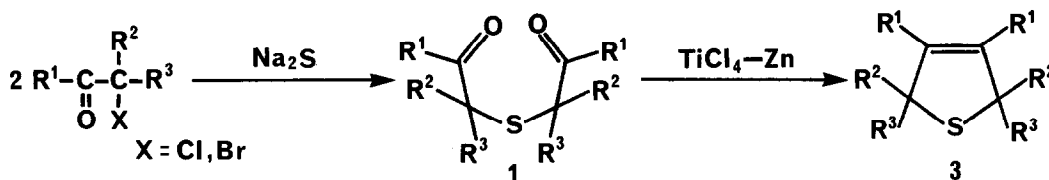


Table 1 Preparation of Symmetrically Substituted 2,5-Dihydrothiophenes (3)

run	R ¹	R ²	R ³	conditions (solvent, temperature, time)	yield (%)
1	C ₆ H ₅	H	H	dioxane, 18 h at r.t.	67 ^{a)}
2	C ₆ H ₅	H	H	dioxane, 4 h at r.t. and then 2 h at 50 °C	62
3	C ₆ H ₅	H	H	THF, reflux, 5 h	77
4	2-naphthyl	H	H	as above	71
5	C ₆ H ₅	CH ₃	H	as above	73 ^{a, b)}
6	C ₆ H ₅	C ₂ H ₅	H	as above	86 ^{b)}
7	2-thienyl	H	H	dioxane, 4 h at r.t. and then 2 h at 50 °C	66
8	4-CH ₃ C ₆ H ₄	H	H	as above	68 ^{a)}
9	4-CH ₃ OC ₆ H ₄	H	H	as above	70
10	4-BrC ₆ H ₄	H	H	as above	68
11	C ₆ H ₅	CH ₃	CH ₃	THF, reflux, 5 h	75

a) These dihydrothiophenes were converted to the corresponding thiophenes quantitatively by oxidation with DDQ in refluxing dioxane. b) The use of the pure meso-sulfides provides *cis*-2,5-dialkyl-3,4-diphenyl-2,5-dihydrothiophenes, exclusively.

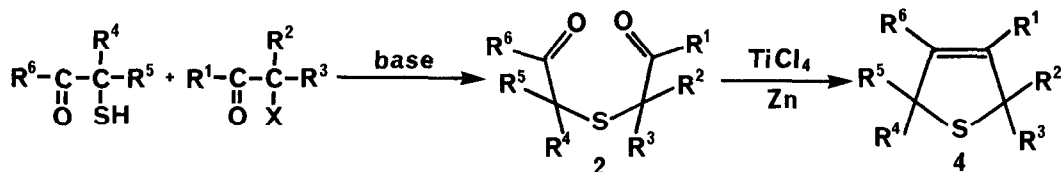



Table 2 Preparation of Unsymmetrically Substituted 2,5-Dihydrothiophenes (4)

run	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	yield (%) of 4 ^a)
1	C ₆ H ₅	H	H	H	CH ₃	CH ₃	82
2	C ₆ H ₅	H	H	CH ₃	CH ₃	CH ₃	81
3	C ₆ H ₅	C ₆ H ₅	H	H	CH ₃	CH ₃	76 ^{b, c)}
4	C ₆ H ₅	CH ₃	H	H	CH ₃	CH ₃	63 ^{b)}
5		(CH ₂) ₂ -	H	H	CH ₃	CH ₃	44 ^{b, d)}

a) THF was used as the solvent. b) The starting diketo sulfides are mixtures of erythro- and threo-isomers and therefore the resulting 2,5-disubstituted-2,5-dihydrothiophenes are mixtures of cis- and trans-isomers. c) Oxidation of this compound with DDQ affords 2,3-dimethyl-4,5-diphenylthiophene. d) Oxidation of this compound by DDQ (2 mol. equiv) leads to 1,2-dimethylnaphtho[2,1-b]thiophene.

Symmetrically substituted diketo sulfides (1) are readily prepared in high yields by reaction of α -haloketones with sodium sulfide. Unsymmetrically substituted diketo sulfides (2) are obtained in excellent yields by reaction of α -haloketones with α -mercaptoketones. We now found that a low-valent titanium reagent, prepared from titanium(IV) chloride and zinc powder in situ,² brings about the intramolecular coupling reaction of 1 and 2 to provide an efficient synthesis of 2,5-dihydrothiophenes (3 and 4).

Tables 1 and 2 summarize preparation of symmetrically substituted 2,5-dihydrothiophenes 3 from 1 and unsymmetrically substituted 2,5-dihydrothiophenes 4 from 2, respectively. These results show that the reaction is general and especially useful for the preparation of highly substituted 2,5-dihydrothiophenes which are otherwise difficult to prepare. Neither products arising from an intermolecular coupling reaction nor isomeric 2,3-dihydrothiophenes were formed.^{3,4}

References and Notes

- (a) J. M. McIntosh and H. B. Goodbrand, *Tetrahedron Lett.*, 3157 (1973); (b) J. M. McIntosh, H. B. Goodbrand, and G. M. Hasse, *J. Org. Chem.*, 39, 202 (1974); (c) J. M. McIntosh and R. S. Steevensz, *Canad. J. Chem.*, 52, 1934 (1974); (d) J. M. McIntosh and G. M. Hasse, *J. Org. Chem.*, 40, 1294 (1975); (e) J. M. McIntosh and R. S. Steevensz, *Canad. J. Chem.*, 55, 2442 (1977); (f) J. M. McIntosh and R. A. Sieler, *J. Org. Chem.*, 43, 4431 (1978); (g) S. F. Birch and D. T. McAllen, *J. Chem. Soc.*, 2556 (1951); (h) G. Stork and P. L. Stotter, *J. Am. Chem. Soc.*, 91, 7780 (1969); (i) R. M. Kellog, S. Wassenaar, and J. Buter, *Tetrahedron Lett.*, 4689 (1970); (j) B. M. Trost and S. D. Ziman, *J. Am. Chem. Soc.*, 93, 3825 (1971); (k) P. L. Stotter, S. A. Roman, and C. L. Edwards, *Tetrahedron Lett.*, 4071 (1972); (l) S. Sadeh and Y. Gaoni, *Tetrahedron Lett.*, 2365 (1973); (m) R. H. Everhardus, R. Grafing, and L. Brandsma, *Recl. Trav. Chim. Pays-Bas*, 95, 153 (1976); (n) G. Field, *J. Org. Chem.*, 44, 825 (1979); (o) K. Kosugi, A. V. Anisimov, H. Yamamoto, R. Yamashiro, K. Shirai, and T. Kumamoto, *Chem. Lett.*, 1341 (1981).
- T. Mukaiyama, T. Sato, and J. Hanna, *Chem. Lett.*, 1041 (1973).
- Some by-products were occasionally isolated in low yields; for example, in run 3 in Table 1, 2,3-diphenyl-1,3-butadiene (trace), 3,4-diphenylthiophene (11%), and 3,4-dihydroxy-3,4-diphenylthiolane (trace) were isolated.
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